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Synthesis and Characterization of Polyphenothiazene Iodine Complexes

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SYNTHESIS AND CHARACTERIZATION OF POLYPHENOTHIAZENE IODINE COMPLEXES

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Abstract. N-methyl 3,6 dibromophenothiazene has been polymerized by using a nickel complex-assisted Grignard coupling. Vapor phase iodine doping of the organometallic polymerized material led to a conductivity increase of 10^{-11} ohm $^{-1}$ cm $^{-1}$ to 10^{-5} ohm $^{-1}$ cm $^{-1}$ at room temperature. The complex thus formed exhibited a semiconductor behavior with a thermal activation energy of 0.1 eV and a strong electronic IR absorbance from 4000-400 cm $^{-1}$. However, the complexed polymer could be dissolved in liquid I_2 and cast into films with air stable, room temperature conductivities as high as $1 \text{ ohm}^{-1}\text{cm}^{-1}$. The mechanisms that could give rise to these differences in behavior will be discussed.